## PRACTICALS (202201)

## **OBJECTIVES : -**

- 1. To conduct chemical analyses by qualitative and quantitative analysis of metal complexes.
- 2. To Perform/demonstrate the techniques involved in organic binary mixture separation specially solid- liquid mixture.
- 3. To interpret the experimental results obtained by potentiometer, pH meter, conductometer.
- 4. To conduct the experiment on various instrumental techniques.
- 5. To describe the principles behind the experiment performed in the laboratory.
- 6. To develop skills in chromatographic techniques for analysis.

Code : 202201	PRACTICAL	4 CREDITS
Inorganic	Semi micro qualitative inorganic analysis:- Identification	8 Hours/ Week
Chemistry	of three acidic and three basic radicals including one rare	
	earth from the given mixture.	
Organic Chemistry	Qualitative Organic Analysis:- Separation, purification and	
	identification of binary (Solid-Liquid) mixtures. The	
	separation should be carried out using Chemical method.	
	The two components are solid-liquid mixtures. Student	
	should submit the purified samples of the separated	
	compounds and prepare a suitable derivative of the two	
	compounds separated out.	
Physical Chemistry	Instrumentation:-	
	1. Determination of strengths of halides in a mixture	
	potentiometrically.	
	2. Determination of the strength of strong and weak acid	
	in a given mixture conductometrically.	
	3. Determination of solubility and solubility product of	
	sparingly soluble salt BaSO4.	
Analytical	Chromatography:- Ion- exchange chromatography, Thin	
Chemistry	layer chromatography.	

## **Reference Books :**

- 1. A Text book of Micro and Semi micro Qualitative Inorganic Analysis, IV edn, A. I. Vogel
- 2. A Text book of Quantitative Inorganic Analysis; A. I. Vogel
- 3. Practical Inorganic Chemistry- Pass Geoffrey and Haydn Sutcliffe.
- 4. Advanced Practical Inorganic Chemistry- Gurudeep Raj.
- 5. Vogel's Qualitative Inorganic Analysis, VII Edn. Orient Longman Ltd. D. Svehla.
- 6. Systematic experimental physical chemistry T. K. Chondhekar & S.W. Rajbhoj
- 7. Experiments in chemistry D.V. Jahagirdar